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Regulating thermal management by a CsPbClBr₂/Ag hybrid microcavity for stable room temperature blue lasing with low threshold

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ABSTRACT

Realization of long-term stable lasing from all-inorganic perovskite supercrystals is highly desirable for practical applications in optoelectronic devices. However, lasing from perovskite supercrystals excited by continuous wave laser light remains a challenge due to the photoluminescence degradation induced by thermal accumulation. Here, we report highly stable lasing with low threshold from a CsPbClBr₂ supercrystal placed on a thin Ag film, which form a CsPbClBr₂/Ag microcavity, by managing the thermal distribution inside the CsPbClBr₂ supercrystal. Combined numerical simulations and lifetime measurements, the localized electric field in such a hybrid microcavity leads to a spatially localized temperature distribution, which plays a crucial role in suppressing the thermal accumulation on the surface and in eliminating non-radiative recombination defects. The effective thermal management in the hybrid microcavity renders highly stable lasing with low threshold under the irradiation of continuous wave laser light. Our findings provide a feasible and universal approach to the development of long-term stable perovskite laser.

1. Introduction

All-inorganic cesium lead halide perovskites (CsPbX₃, (X=Cl, Br, I)) supercrystals (SCs), which are formed by the self-assembly of CsPbX₃ quantum dots (QDs)[1–4], have attracted great interest due to their outstanding physical and chemical properties, such as superfluorescence (SF)[2,5,6], long-term stability[7], and high optical gain[8,9]. Recently, in order to overcome optical loss and bring about new functions, optical resonant nanostructures have been introduced and extensively studied over the last decade[10–12]. In particular, CsPbX₃ SCs with micrometer sizes can serve as optical microcavities supporting whispering gallery modes (WGMs) owing to their regular geometric morphologies and smooth surfaces[13–15]. For this reason, stimulated emission from CsPbX₃ SCs can be easily realized. For instance, room temperature nanosecond-sustained green lasing with a low threshold (~25 μJ/cm²)

was demonstrated very recently in CsPbBr₃ SCs[16]. However, one of the main obstacles that limits the lasing stability of mixed halide CsPbX₃ SCs is light-induced migration of halide anion mediated by defect states, which leads to the phase segregation[17–19]. In addition, the thermal accumulation on the surfaces of CsPbX₃ SCs usually causes local overheating under the irradiation of continuous wave laser light due to the low thermal conductivity of the perovskite (~0.4 W/m•K), leading to the degradation of the photoluminescence (PL) [20,21]. This surface thermal accumulation becomes more serious for CsPbX₃ SCs with micrometer sizes, dramatically degrading the emission efficiencies of CsPbX₃ SCs. So far, the lasing stability of CsPbX₃ SCs hinders the practical application in laser devices.

Organic ligands decorated on the surfaces of perovskite QDs, such as polymer compounds and oleic acid, can increase inter-particle spacing and prevent the aggregation of perovskite QDs, making them stable and

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monodisperse in solution [22,23]. On the other hand, the interaction between the surface ligands of adjacent QDs can be exploited to realize the assembly of QDs. It has been demonstrated that CsPbX₃ SCs can be fabricated via polar solvent-assisted self-assembly of QDs by balancing the interaction force between solution–ligand and that between ligand–ligand [24–27]. In this case, the detachment of surface ligands induced by polar solvent will create dangling bonds on the surfaces of QDs. Such dangling bonds would induce local defects or structural distortions in CsPbX₃ SCs, which are responsible for the thermal-activated nonradiative processes and high Auger recombination losses for CsPbX₃ SCs under high excitation fluences [28–30]. The existence of such defects hampers the achievement of low-threshold lasing in CsPbX₃ SCs. In addition, CsPbX₃ SCs often suffer from significant compressive strain due to the soft nature of ligand shells on QDs, which is detrimental to the stability of SCs. Therefore, fabrication of high-quality CsPbX₃ SCs while suppression of laser-induced overheating is crucial for realizing stable lasing of CsPbX₃ SCs at room temperature.

Laser annealing is a common method used for eliminating the defects and residual strain in a crystal, including CsPbX₃ microplatelets [31–38]. Due to the low thermal conductivity of CsPbX₃, however, the thermal management in a CsPbX₃ SC during laser annealing is necessary in order to avoid the thermal accumulation on the surface of the CsPbX₃ SCs. Physically, the temperature distribution in a microcavity is strongly correlated with the electric field distribution in the microcavity, which determines the energy distribution of the irradiation laser light. It has been known that a hybrid nanocavity formed by a dielectric nanoparticle and a thin metal film can be employed to establish a spatially localized temperature distribution inside the nanocavity [32,39]. In addition, the thin metal film with a good thermal conductivity acts as a heat sink, effectively suppressing the thermal accumulation in the nanocavity [40–44]. Therefore, it is expected that the hybrid microcavity formed by a CsPbX₃ SC and a thin metal film can enhance the interaction of the CsPbX₃ SC with laser light and prevent the damage of CsPbX₃ SC from overheating. In this way, the long-term stability of CsPbClBr₂ SCs can be achieved under the irradiation of continuous wave laser light, which is crucial for realizing stable lasing at room temperature.

In this work, we investigated the thermal management in a hybrid microcavity formed by placing a CsPbClBr₂ SC on a thin silver (Ag) film. We found that the electric field distribution in the hybrid microcavity leads to a uniform temperature distribution in the CsPbClBr₂ SC, which is helpful for suppressing the thermal accumulation on the surface and reducing the non-radiative recombination of the defects. We demonstrated the stable lasing of the CsPbClBr₂/Ag with a low threshold under the excitation of continuous wave laser light at room temperature.

2. Materials and methods

2.1. Fabrication of CsPbClBr₂ QDs

In this work, CsPbClBr₂ QDs were synthesized by hot-injection method. Cesium carbonate (Cs₂CO₃, Aladdin, 99.99 %), lead(II) bromide (PbBr₂, Aladdin, 99.99 %), lead(II) chloride (PbCl₂, Aladdin, 99.99 %), oleic acid (OA, Aladdin, 90 %), oleyamine (OAm, Aladdin, 90 %), trioctylphosphine (TOP, Aladdin, 90 %) and octadecene (ODE, Aladdin, 95 %) were used as the reactant precursors. First, Cs-oleate precursors were fabricated according to a typical synthesis. 100 mg Cs₂CO₃, 1 mL OA were mixed with 10 mL ODE in a 100 mL three-neck flask. 138 mg PbBr₂, 103 mg PbCl₂, 2 mL TOP, 2 mL OA, and 2 mL OAm were mixed with 20 mL ODE in another 100 mL three-neck flask. Then, the above solutions were maintained at 90 °C for 30 min under vacuum in order to remove air and moisture and then heated to 120 °C for 60 min until the Cs-oleate/Pb-oleate is completely dissolved in ODE. After that, the temperature of Cs-oleate solution was kept at 120 °C under nitrogen atmosphere for further hot injection, while the temperature of Pb-oleate solution was gradually increased to 180 °C under nitrogen atmosphere and kept for 20 min. Subsequently, 2 mL hot Cs-oleate solution was

swiftly injected into the Pb-oleate solution. After 5 s, the reaction mixture was immediately cooled down in an ice-water bath. The crude CsPbClBr₂ QDs solution was centrifuged at 9200 rpm for 5 min. The concentrated supernatant was dispersed with a mixture of hexane and acetone with a volume ratio of 1:1 and then centrifuged at 8800 rpm for 5 min. The final CsPbClBr₂ QDs supernatant was dissolved in the hexane solution.

2.2. Synthesis of CsPbClBr₂SC/Ag hybrid microcavities

The as-prepared 4 mL CsPbClBr₂ QDs solution added 4 mL methyl acetate was left standing at room temperature for ten days. Then 50 µL solution was spin-coated onto a 10*10 mm clean Ag/SiO₂ substrate at a rate of 4000 rpm for 30 s. The CsPbClBr₂ SC were form on the surface of Ag film. The 50-nm-thick Ag film used in this work was prepared on a SiO₂ substrate via thermal evaporation method, and the spatial distribution of Ag is uniform. After that, the substrates were dried in a vacuum oven for 12 h in ambient condition.

2.3. Characterization of CsPbClBr₂ QDs and CsPbClBr₂ SC/Ag hybrid microcavities

The morphology and component elements of CsPbClBr₂ SC were characterized by a scanning electron microscope (SEM) (Hitachi SU5000), a high-resolution transmission electron microscope (HR-TEM) (JEM-1400 PLUS) and energy dispersive spectroscopy (EDS) (Bruker EDS QUANTAX), respectively. The crystal structures of CsPbClBr₂ QDs and CsPbClBr₂ SC were characterized by X-ray diffraction (XRD) (Bruker D8 Advance) at 35 kV and 35 mA. The UV–Vis absorption spectra were measured by using a spectrophotometer (Shimadzu UV–Vis 3600). The PL spectra of CsPbClBr₂SC were collected by a 40 × UV objective lens in a Raman spectrometer system (Horiba LabRAM HR Evolution) equipped with a 325-nm He-Cd laser (Kimmon). The PL spectra of CsPbClBr₂ SC were collected by a 40 × UV objective lens in a Raman spectrometer system (Horiba LabRAM HR Evolution) equipped with a 325-nm He-Cd laser (Kimmon) and a long wave pass edge 325 nm-filter.

2.4. Optical characterization of CsPbClBr₂ SC/Ag hybrid microcavities

The 800 nm femtosecond laser light (Mira 900S, Coherent, 130 fs) was employed to excite CsPbClBr₂ SC, a high repetition rate 76 MHz was measure luminescence and power-dependent, and a low repetition rate of 3.8 MHz (obtained by using a pulse picker) and a time-correlated single photon system (lifespec II, Edinburgh Instruments) was employed to measure luminescence lifetimes. The single-mode lasing spectra measurement were performed by using 400 nm femtosecond pulses of 1 kHz (Legend, Coherent). A 325 nm CW laser was measured multimode lasing spectra. The laser light was focused on the samples by using the 50× objective lens (NA=0.8) of an inverted microscope (Axio Observer A1, Zeiss). The scattering light and photoluminescence were collected by using the same objective lens and directed to a spectrometer (SR-500i-B1, Andor) for analysis or to a coupled-charge device (DU970N, Andor) for recording.

2.5. Numerical simulation and analytical model

The scattering spectra of CsPbClBr₂ SC placed on a thin Ag film were calculated numerically by using the finite-difference time-domain (FDTD) method (FDTD solution, <https://www.lumerical.com>). In the numerical simulation, the side length of the CsPbClBr₂ SC is 2 × 2 µm and the height is 0.5 µm and is placed on the surface of Ag film. The dielectric constants of Ag was taken from the previous literature [45,46], and refraction index of CsPbClBr₂ SC were taken from the previous literature [47,48] with fitting the absorption peak in the measured spectra in our data. The dipole source was placed inside the CsPbClBr₂ SC to calculate the radiation intensity of the CsPbClBr₂/SiO₂

and CsPbClBr₂/Ag. In order to obtain converged simulation results, a perfectly matched layer boundary condition was employed to terminate the finite simulation region. The temperature distributions inside the hybrid microcavities were calculated numerically based on the finite element method (FEM) (COMSOL Multiphysics v5.6, <https://www.comsol.com>). The refractive index and thermal conductivity of CsPbClBr₂ SC were taken from literature[48–50].

3. Results and discussion

As shown in Fig. 1(a), we synthesized CsPbClBr₂ SCs with regular geometric shape by the self-assembly of monodispersed QDs assisted by polar solvent. The CsPbClBr₂ QDs were prepared by using typical high temperature hot injection following the procedures described in literature. Based on the transmission electron microscopy (TEM) image (Fig. 1(b)), the average size of the CsPbClBr₂ QDs was estimated to be $\sim 10 \pm 2$

nm. In addition, the distance between the lattice fringes of QDs was found to be ~ 0.29 nm, corresponding to the (001) crystal plane of orthorhombic CsPbClBr₂[51]. This assignment was further confirmed by the X-ray diffraction (XRD) measurements shown in Fig. 1(c), where two double sharp peaks corresponding to the orthorhombic phase of CsPbClBr₂ are observed at $\sim 15.7^\circ$ and $\sim 31.4^\circ$. In Fig. 1(d), we present the PL spectrum of CsPbClBr₂ QDs, which appears as a highly symmetric emission band at ~ 478 nm with a full width at half maximum (FWHM) of ~ 21 nm. All these characterizations indicate the high crystalline quality of the CsPbClBr₂ QDs used in the fabrication of CsPbClBr₂ SCs.

We also examined the crystal structures, morphologies and optical properties of the synthesized CsPbClBr₂ SCs. As shown in Fig. 1(c), the XRD spectrum of the CsPbClBr₂ SCs remains unchanged as compared with that of the CsPbClBr₂ QDs. The morphologies of the CsPbClBr₂ SCs were characterized by using scanning electron microscopy (SEM). The SEM image of a typical Fig. 1(e). It can be seen that the CsPbClBr₂ SC

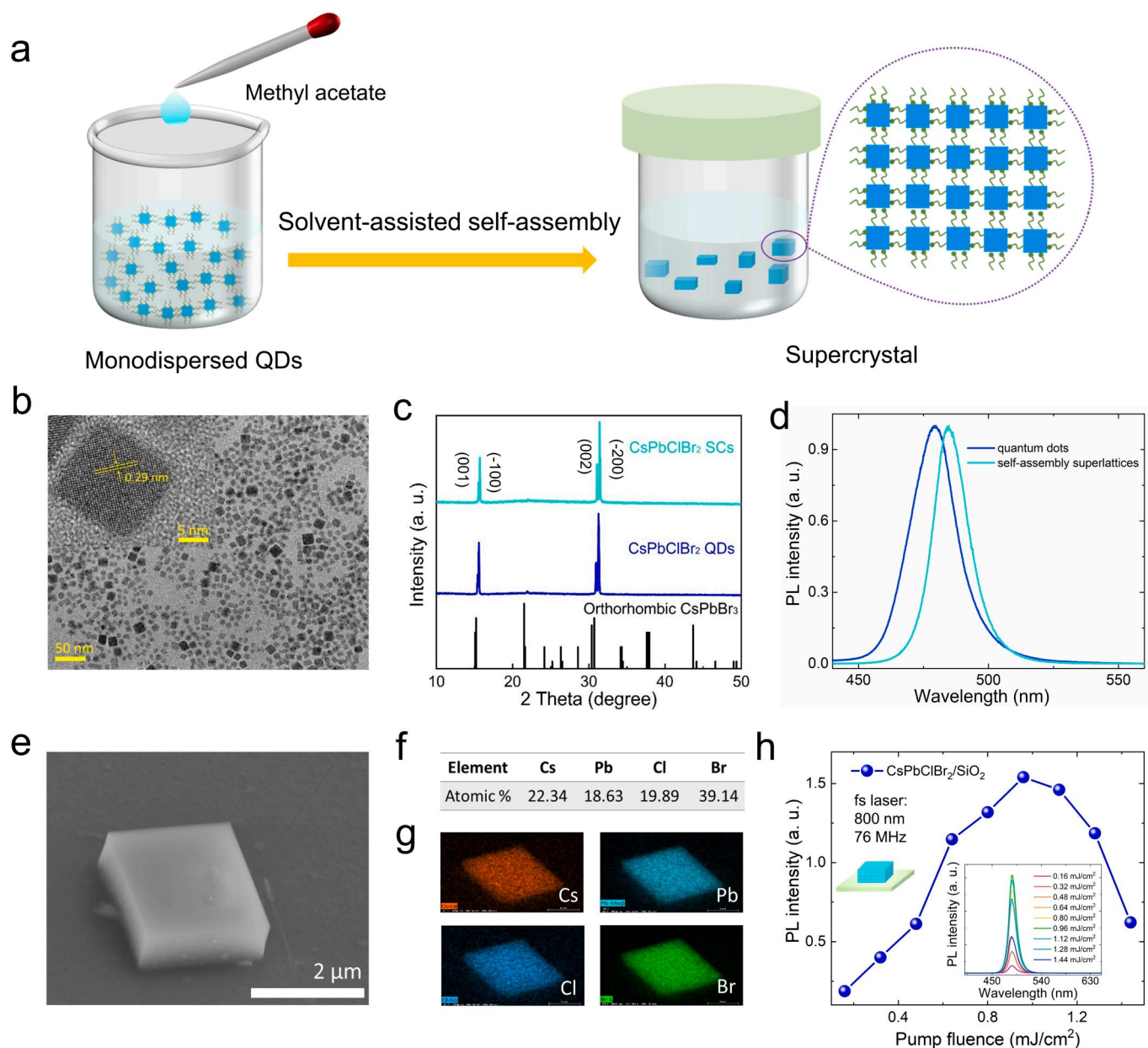


Fig. 1. (a) Schematics of the fabrication process of CsPbClBr₂ SC. (b) TEM image of the pristine CsPbClBr₂ SC. (c) XRD patterns of the CsPbClBr₂ SC, CsPbClBr₂ QDs, and reference peak of the CsPbBr₃. (d) The PL spectra of CsPbClBr₂ QDs and CsPbClBr₂ SC. (e) SEM image of a CsPbClBr₂ SC on an Ag film surface (f, g) EDS analysis of Cs, Pb, Cl, Br. (h) PL spectra measured for a CsPbClBr₂ supercrystal placed on the SiO₂ substrate excited by using 800 nm femtosecond laser pulses of 76 MHz at different pump fluence.

appears as a cuboid with a flat and smooth surface. The elemental analysis and mapping based on energy dispersive spectroscopy (EDS) are shown in Fig. 1(f) and (g), verifying the uniform distributions of elements (Cs, Pb, Cl, and Br) in the CsPbClBr₂ SC. The atomic ratio of Cs/Pb/Cl/Br = 1:1:1:2 is in good agreement with the stoichiometric ratio of CsPbClBr₂. Moreover, the PL spectrum of the CsPbClBr₂ SC is different from that of CsPbClBr₂ QDs, as shown in Fig. 1(d). A redshift of the PL peak (from ~ 478 to ~ 485 nm) and a narrowing of the FWHM (from ~ 21 to ~ 14 nm) is observed for the CsPbClBr₂ SC. This change in the PL spectrum is attributed to the formation of miniband in the CsPbClBr₂ SC due to the electronic coupling of neighboring QDs[52]. Remarkably, the emission from the CsPbClBr₂ SC remains stable for more than 30 min under the excitation of 325-nm continuous wave (CW) laser light with a power density of $P=44.71 \text{ W/cm}^2$. This behavior is in contrast to the PL degradation observed for the QDs after only 3 min under the same excitation condition. Moreover, no halide phase segregation was observed for the CsPbClBr₂ SC. It indicates the remarkably improved photo-stability in the SC formed by the assembly of QDs, as reported in a previous study[53]. This suppression of halide phase separation is attributed to the reduction of exciton-phonon coupling and lattice strain resulting from the enhanced excitonic delocalization in the CsPbClBr₂ SC. Therefore, assembled superstructures with new properties generated by electronic coupling between QDs are expected to be applied in perovskite laser devices.

In fact, local defects and/or structural distortions are inevitably introduced in the fabrication process of CsPbClBr₂ SCs due to the

detachment of surface ligands or large lattice mismatch. Although the photo-stability of CsPbClBr₂ SCs can be significantly improved as compared with QDs, PL degradation still occurs in a CsPbClBr₂ SC excited by using a high power density. In Fig. 1(h), we show the evolution of the PL intensity with increasing laser pump fluence observed for a CsPbClBr₂ SC placed on a SiO₂ substrate by using 800-nm femto-second laser pulses of 76 MHz. A rapid PL quenching is clearly observed for laser fluence larger than 0.96 mJ/cm^2 . Due to the low thermal conductivity of the perovskite, the local overheating induced by laser irradiation in the CsPbClBr₂ SC may lead to the aggregation of QDs, increasing surface defects and quenching the PL.

The recent results suggest that perovskites SC can be empowered by optical resonances, and the tight confinement of the electromagnetic field within it can promote light-matter interactions and control light emission at the wavelength of the material emission[6,12,53]. For the hybrid microcavity created by placing a CsPbClBr₂ SC on a thin Ag film, it is necessary to identify the optical modes supported by the hybrid microcavity, which can be employed to enhance the linear and nonlinear optical responses of the CsPbClBr₂ SC. In Fig. 2(a), we show the absorption and PL spectra of the CsPbClBr₂ SC. It was found that absorption and PL peaks of the CsPbClBr₂ SC was located at 492 nm. The refractive index of CsPbClBr₂ SC was taken from the previous literature [47,48] with fitting the absorption peak in the measured spectra in our data (see Fig. 2(b)). In Fig. 2(c), we show the radiation spectra of two CsPbClBr₂ SCs placed on an Ag/SiO₂ substrate and a SiO₂ substrate, respectively. In each case, a dipole source placed at the center of the

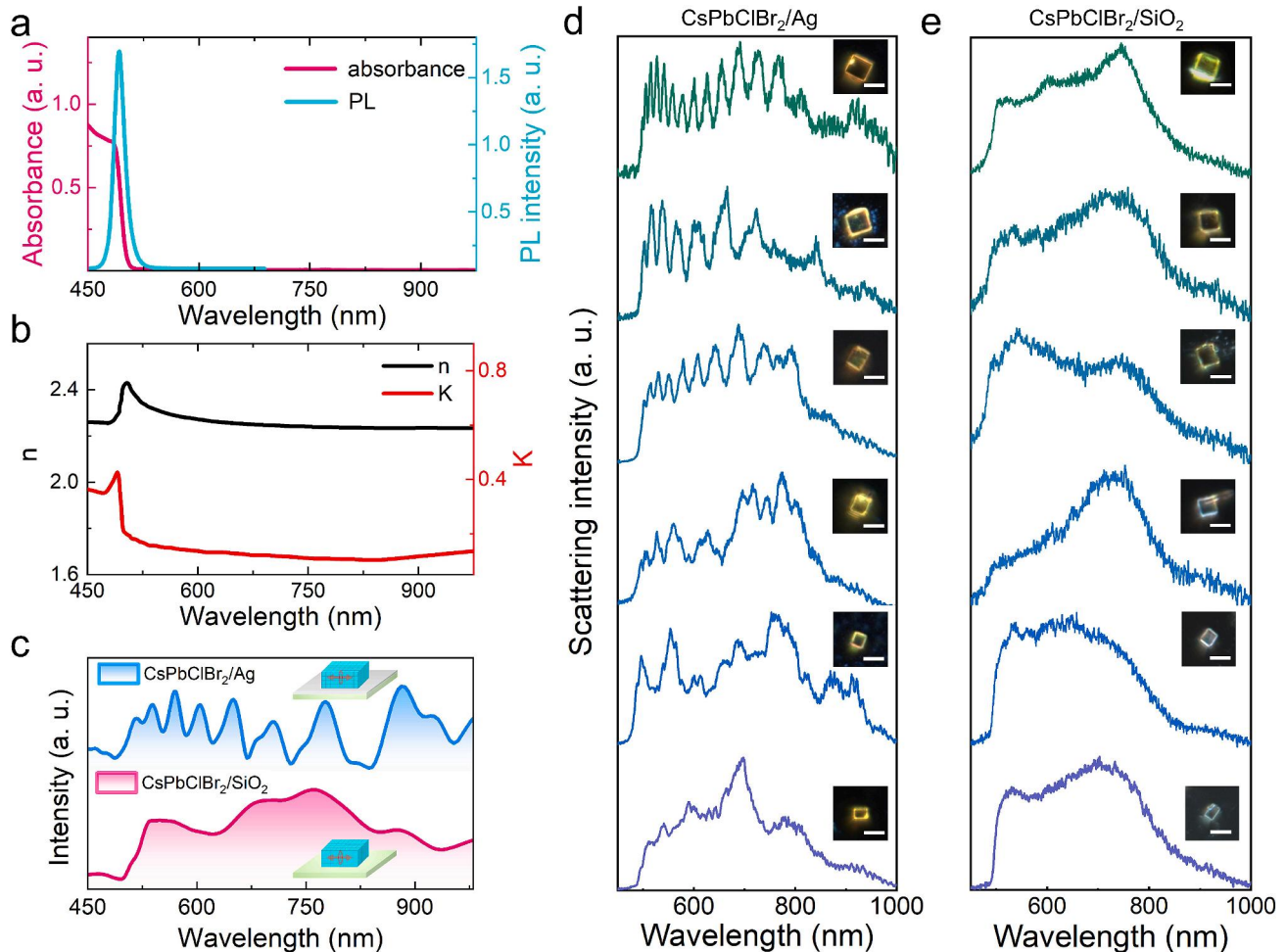


Fig. 2. (a) The absorption and PL spectra of CsPbClBr₂ SC. (b) The refractive index of CsPbClBr₂ SC. (c) Radiation spectra calculated for a CsPbClBr₂ SC placed on the Ag/SiO₂ and SiO₂ substrates, respectively. The side length and thickness of the cuboidal SC are assumed to be 2 μm and 0.5 μm . Scattering spectra measured for CsPbClBr₂ SCs with different sizes placed on an Ag/SiO₂ substrate (d) and a SiO₂ substrate (e). Scale bar: 2 μm .

CsPbClBr₂ SC was used as the excitation source. It noticed that the CsPbClBr₂ SC placed on the Ag/SiO₂ substrate supports a variety of optical modes with high quality (Q) factors. In comparison, very few optical modes with large Q factors are observed in the CsPbClBr₂ SC placed on the SiO₂ substrate. For the hybrid microcavity composed of the CsPbClBr₂ SC and the Ag film, the high-Q optical modes may originate from the coherent interaction between the high-order Mie resonances and the whispering gallery modes (WGM) or Fabry-Perot (F-P) resonances. They are beneficial for the light amplification in the CsPbClBr₂/Ag hybrid microcavity. In order to gain a deep insight into

the optical modes supported by CsPbClBr₂/Ag hybrid microcavities, we measured the scattering spectra of CsPbClBr₂/Ag hybrid microcavities with different sizes by using polarized white light and analyzer with cross polarization, as shown in Fig. 2(d). The scattering spectra measured for CsPbClBr₂ SCs placed on the SiO₂ substrate are also provided for comparison (see Fig. 2(e)). One can identify high-Q optical modes at the emission wavelength of CsPbClBr₂, especially for hybrid microcavities with large sizes. In comparison, very few optical modes with high Q factors are revealed in the CsPbClBr₂ SCs placed on the SiO₂ substrate. The Q value of most of the measured resonance mode is close

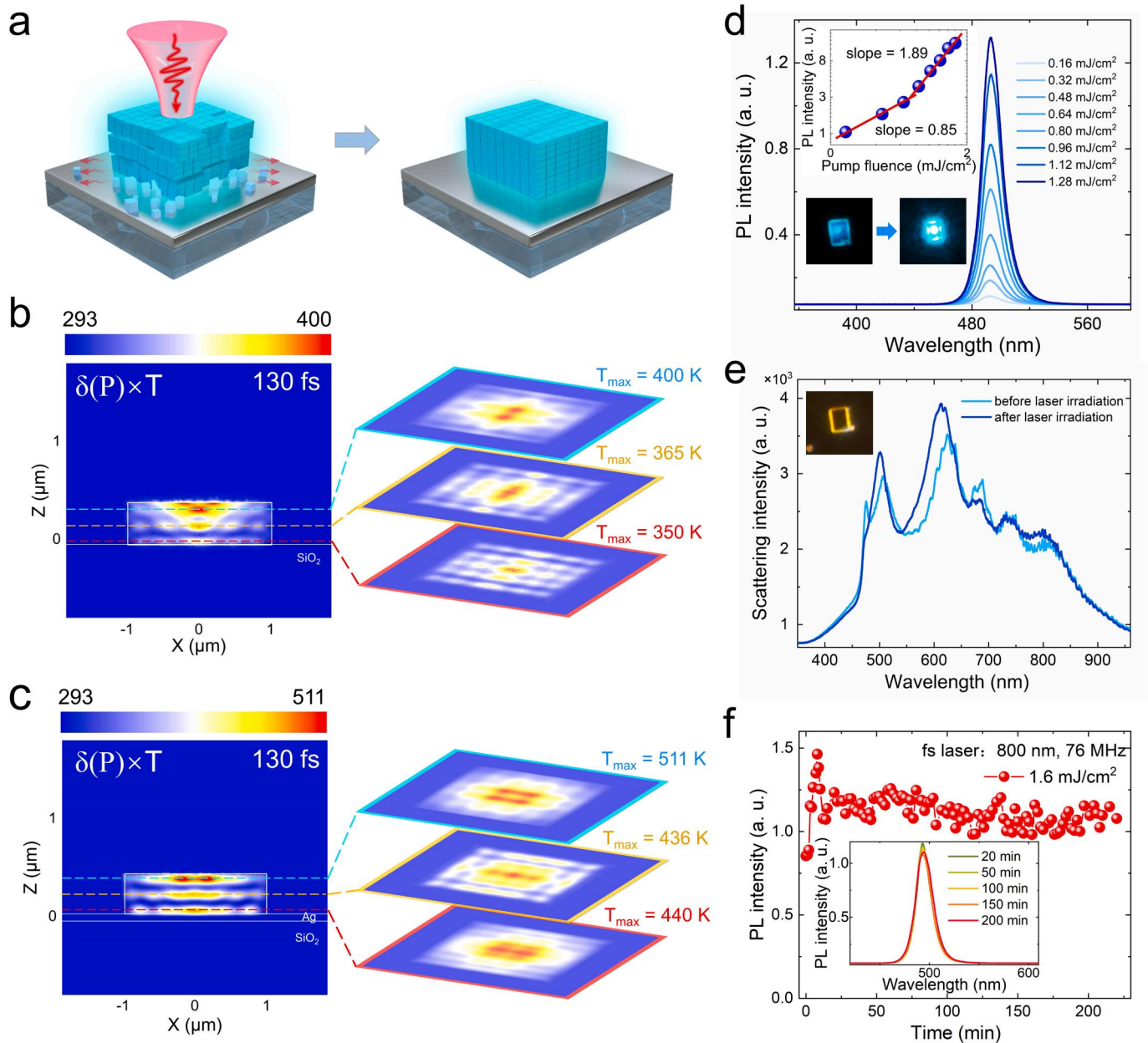


Fig. 3. (a) Schematic showing the annealing of a CsPbClBr₂ SC placed on a Ag/SiO₂ substrate induced by laser irradiation. (b) Transient temperature distribution in the XZ and XY planes ($t = 130$ fs) calculated for a CsPbClBr₂ SC placed on a SiO₂ substrate and excited by using a single 800-nm femtosecond laser pulse with a duration of 130 fs. (c) Transient temperature distribution in the XZ and XY planes ($t = 130$ fs) calculated for a CsPbClBr₂ SC placed on an Ag/SiO₂ substrate and excited by using a single 800-nm femtosecond laser pulse with a duration of 130 fs. (d) PL spectra measured for a CsPbClBr₂/Ag hybrid microcavity excited by using 800-nm femtosecond laser pulses at different pumping fluences. The dependence of the PL intensity on the pumping fluence plotted in a logarithmic coordinate and the optical image of the CsPbClBr₂ SC recorded by using a CCD at different pumping fluences are provided in the insets. (e) Forward scattering spectra measured for the CsPbClBr₂/Ag hybrid microcavity before and after the laser irradiation. The optical image of the CsPbClBr₂ SC recorded by using a CCD is shown in the inset. (f) PL intensities of the CsPbClBr₂/Ag hybrid microcavity excited continuously by using 800-nm femtosecond laser pulses. The PL spectra recorded at different times are shown in the inset.

to the simulation, and only some resonance mode has lower Q value. This is mainly because compared with the simulation model, the synthesized CsPbClBr₂ SC edge has a certain curvature, which is not so sharp.

As demonstrated above, the optical modes supported by a CsPbClBr₂/Ag hybrid microcavity possess electric field distributions similar to standing waves inside the CsPbClBr₂ SC. At the excitation wavelength, such a localized electric field distribution leads to a spatially-localized temperature distribution, which is helpful for eliminating the defects induced in the CsPbClBr₂ SC during the self-assembly process, as schematically illustrated in Fig. 3(a). As a result, the CsPbClBr₂/Ag hybrid microcavity is expected to exhibit improved stability under the excitation of laser light. In order to find out the effect of the Ag film on the temperature distribution in a CsPbClBr₂ SC, we employed numerical simulation to calculate the temperature change induced in a CsPbClBr₂ SC placed on a SiO₂ substrate by a single 800-nm femtosecond laser pulse with a duration of 130 fs, as shown in Fig. 3(b). The simulation is based on the two temperature model with considering the pulse duration of the pulse and it is widely used in nanophotonic system under femtosecond laser excitation[54–56]. It should be noted that the model is based on a linear model rather than the nonlinear effect such as two/multi-photon absorption for simplicity, which is enough for explaining the laser heating effect on the sample[57,58]. It is found that the temperature rise is not uniform in the SC and it primarily occurs at the top of the SC. This behavior implies that overheating may appear on the surface of the SC, leading to the increase of surface defects. The temperature is gradually transmitted from the top to the bottom, and gradually decreases, only a small region of the CsPbClBr₂ SC is affected by laser heat. This prediction is supported by the PL lifetime measured for the CsPbClBr₂ SC placed on a SiO₂ substrate under continuous laser irradiation (see Fig. 4(a)). In this case, a gradual decrease of the PL lifetime is observed with increasing irradiation time, indicating an increase of nonradiative decay rate. This explains the rapid PL quenching observed for CsPbClBr₂ SCs placed on a SiO₂ substrate at high excitation power densities. Therefore, the thermal management in a CsPbClBr₂ SC plays a crucial role in improving the photo-stability of the CsPbClBr₂ SC.

In Fig. 3(c), we show the temperature change induced in a CsPbClBr₂/Ag hybrid microcavity under the same excitation condition. It is noticed that the temperature rise in the CsPbClBr₂ SC no longer contracts at the top of the SC. Instead, it is uniformly distributed at the top, in the middle and at the bottom of the SC. This modification in the temperature distribution induced by the addition of the Ag film greatly

suppresses the overheating appearing on the surface of the SC under laser irradiation. The temperature of CsPbClBr₂ SC placed on Ag film after laser action is higher than that placed on SiO₂ substrates. Surprisingly, the temperature at the bottom of CsPbClBr₂ SC is actually higher than that in the central region, and the distribution of the high-temperature region is larger, which is caused by the field localization of plasmon at the gap position of CsPbClBr₂ SC and Ag film. This main that more areas are affected by the laser annealing. On the other hand, it is helpful for eliminating the defects induced in the SC during the self-assembly process by exploiting the laser-induced annealing. In order to confirm this suspect, we monitored the PL lifetime of a CsPbClBr₂ SC, as shown in Fig. 4(b). A gradual increase in the PL lifetime is observed with increasing irradiation time, implying the decrease of nonradioactive recombination centers upon the laser irradiation[59]. This behavior indicates the possibility for realizing ultra-stable light emission from a CsPbClBr₂ SC/Ag hybrid microcavity under continuous laser irradiation.

We examined the PL spectra of a CsPbClBr₂ SC/Ag hybrid microcavity by using 800-nm femtosecond laser pulses at different pumping fluences (p), as shown in Fig. 3(d). The dependence of the PL intensity on the pumping fluence plotted in a logarithmic coordinate is provided in the inset. At low pumping fluences, the slope extracted from the dependence of the PL intensity on the pumping fluence (~ 0.82) is smaller than the expected value for two-photon-induced luminescence (TPL) (~ 2.0). A slope of ~ 2.0 is observed at high pumping fluences, indicating the TPL nature of the emission from the hybrid microcavity. The increase of the slope from ~ 0.82 to ~ 2.0 is mainly caused by the elimination of the defects inside the CsPbClBr₂ SC induced by laser annealing. Interestingly, the blue light emission from the centers of four edges can be clearly observed and is considered to result from the WGM modified emission. In order to confirm the effect of laser annealing, we examined the scattering spectra of the CsPbClBr₂/Ag hybrid microcavity before and after the laser irradiation by using dark-field microscopy, as shown in Fig. 3(e). The optical image of the CsPbClBr₂ SC recorded by using a charge coupled device (CCD) is shown in the inset. Surprisingly, we observed increased intensities and blue shifts for the main scattering peaks. It implies that the local defects or structural distortions in the CsPbClBr₂ SC are removed by the temperature rise induced by laser irradiation. As a result, the emission from the CsPbClBr₂ SC becomes more efficient and stable. In Fig. 3(f), we show the PL intensities of the CsPbClBr₂/Ag hybrid microcavity irradiated continuously by using 800-nm femtosecond laser pulses of 76 MHz at a pumping fluence of 1.6 mJ/cm². It can be seen that the PL intensity remained stable for more than

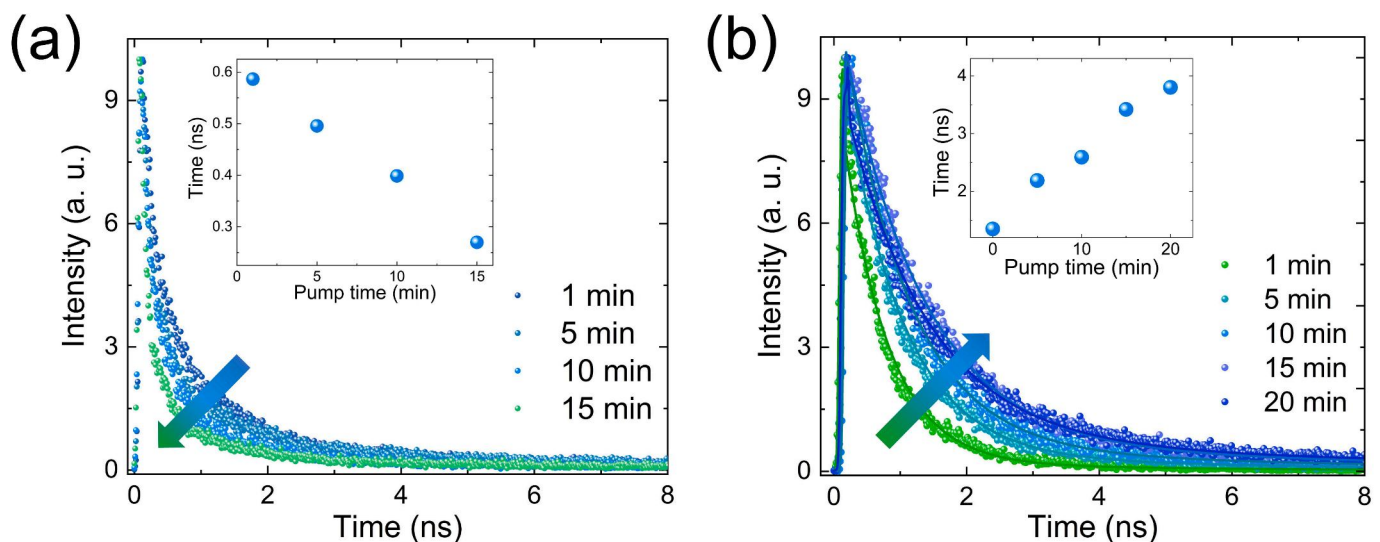


Fig. 4. (a) Pump fluence time-dependent PL decay of CsPbClBr₂ SC place on the SiO₂ substrate at room temperature by using the 800-nm femtosecond laser pulses of 3.8 MHz. (b) Pump fluence time-dependent PL decay of CsPbClBr₂ SC place on the Ag film by using the 800-nm femtosecond laser pulses of 3.8 MHz.

260 min.

At present, the major obstacle for realizing CW-laser-pumped perovskite lasers at room temperature is the thermal accumulation effect induced by CW laser light. As demonstrated above, the hybrid microcavity composed of a CsPbClBr₂ SC and a thin Ag film can effectively suppress the local overheating induced by laser irradiation. Therefore, it is expected that the stimulated emission from such a hybrid microcavity can be achieved under the excitation of CW laser light. In Fig. 5(a), we show PL spectra measured for a CsPbClBr₂/Ag hybrid microcavity at different power densities by using 325-nm CW laser light. One can identify optical modes with narrow linewidths supported by the CsPbClBr₂ SC in the PL spectra. The dependence of the integrated PL intensity and the linewidth (FWHM) on the excitation power density is plotted in Fig. 5(b). It is noticed that the PL intensity increases rapidly when power exceeds a critical value ($P \sim 10 \text{ W/cm}^2$), where a transition from sublinear to superlinear behavior is observed. When the pump density was further increased, the narrow linewidth peaks dominated the emission spectrum and the output intensity increases dramatically, as shown in the inset of Fig. 5(b). Due to the coupling between the PL spectra of CsPbClBr₂ excitons and the WGM supported by CsPbClBr₂/Ag hybrid microcavity, the PL spectra acquired by the hybrid microcavity system shows a narrow linewidth resonance mode. However, the WGM supported by CsPbClBr₂/Ag hybrid microcavity has a high Q factor (as shown in Fig. 2(d)), its PL spectra is not much different from the FWHM of laser, and it is difficult to observe the obvious linewidth narrowing phenomenon at the threshold. In addition, it is found that the linewidth

(FWHM) of the sharp optical mode appearing at $\sim 498 \text{ nm}$ is reduced to $\sim 1.0 \text{ nm}$ when the excitation power density is raised to $P = 44.71 \text{ W/cm}^2$, as shown in the inset of Fig. 5(b). The rapid increase in the PL intensity and the dramatic narrowing of the linewidth above the threshold indicates clearly the multi-mode lasing realized in the CsPbClBr₂/Ag hybrid microcavity. To further evaluate the superiority of the CsPbClBr₂/Ag hybrid microcavity, we tested amplified spontaneous emission (ASE) property by using 400-nm femtosecond laser pumping (130 fs, 1 kHz) at ambient conditions. In Fig. 5(c), we present power-dependent PL spectra of the CsPbClBr₂/Ag hybrid microcavity at different power densities. It is found that under low power densities ($< 36 \mu\text{J/cm}^2$), the intensity of photoemission increased slowly and the PL spectra can see a broad emission band (FWHM $\sim 26.3 \text{ nm}$) centered at $\sim 490 \text{ nm}$. Strikingly, with the excitation power density is increased to $\sim 54 \mu\text{J/cm}^2$, a sharp peak with a narrow linewidth (FWHM $\sim 5.4 \text{ nm}$) emerges at the low-energy shoulder of the emission band ($\sim 498 \text{ nm}$), as shown in Fig. 5(d). The nonlinear response of the light emission intensity and the dramatic narrowing in the linewidth to pump fluence confirms the ASE behaviors in the CsPbClBr₂/Ag hybrid microcavity.

An essential characteristic of optical gain materials is their photostability under laser irradiation. In order to examine the stability of the stimulated emission from the CsPbClBr₂/Ag hybrid microcavity by using 400-nm femtosecond laser pulse (130 fs, 1 kHz), the PL spectra recorded at different times are shown in Fig. 6(a). We monitored the peak intensity of the stimulated emission for more than 60 min, as shown in Fig. 6(b). It can be seen that the PL intensity remains stable for a long

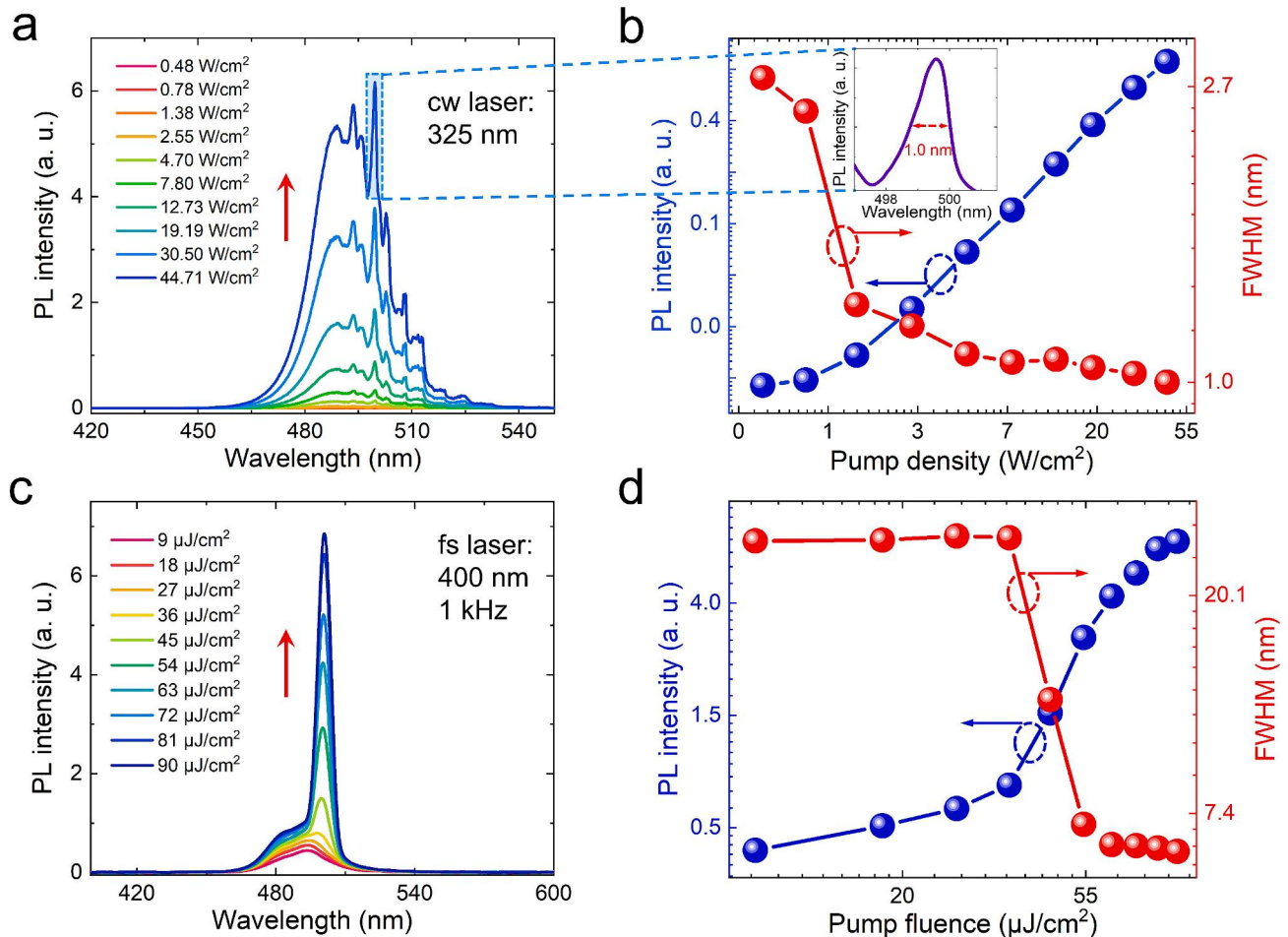


Fig. 5. (a) PL spectra measured for a CsPbClBr₂/Ag hybrid microcavity at different excitation power densities by using 325-nm CW laser light. (b) Dependence of the integrated PL intensity and linewidth on the excitation power density. The inset shows the magnified PL spectrum of a sharp optical mode with a narrow linewidth. (c) PL spectra measured for a CsPbClBr₂/Ag hybrid microcavity at different excitation power densities by using 400-nm femtosecond laser pulses. (d) Evolutions of the PL intensity and linewidth with increasing power density.

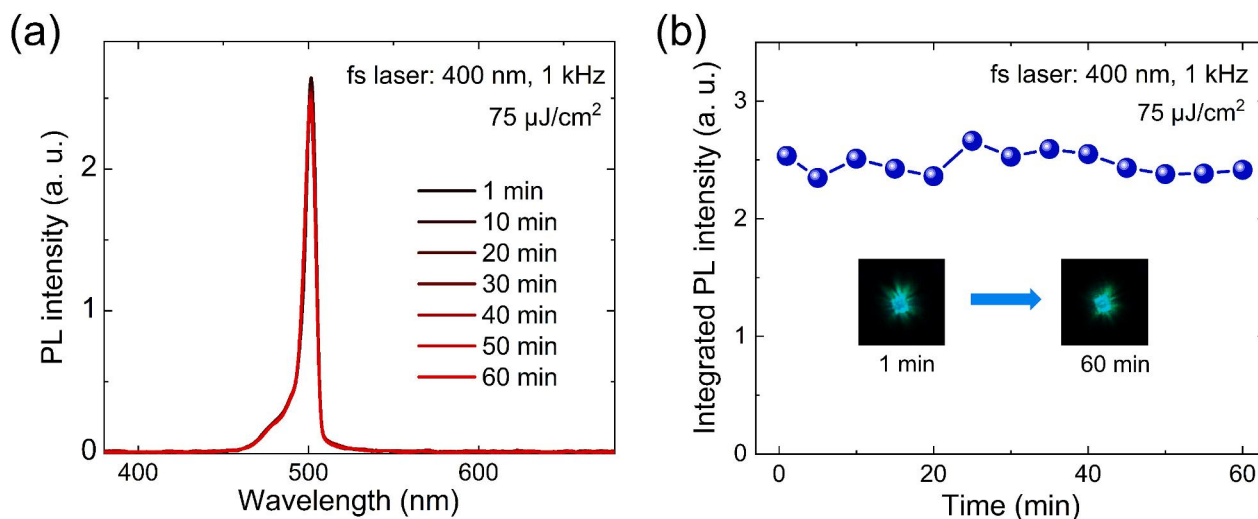


Fig. 6. (a) PL spectrum measured for the CsPbClBr₂/Ag hybrid microcavity at different times under the excitation of 400-nm femtosecond laser pulses. (b) Integrated PL intensities of the CsPbClBr₂/Ag hybrid microcavity recorded at different times under the excitation of 400-nm femtosecond laser pulses. The optical image of the CsPbClBr₂ SC recorded by using a CCD at 1 min and 60 min respectively are shown in the inset.

time under the continuous excitation of 400-nm femtosecond laser pulses. This feature indicates the great potential of CsPbClBr₂/Ag hybrid microcavities in making multimode lasers with low threshold and high stability.

4. Conclusions

In summary, we developed a method to manage the temperature distribution inside a CsPbClBr₂ SC creating a CsPbClBr₂/Ag hybrid microcavity. The strongly localized electric field achieved in the hybrid microcavity leads to relatively uniform temperature distribution inside the CsPbClBr₂ SC, which suppresses the thermal accumulation on the surface of the CsPbClBr₂ SC and reduces the non-radiative recombination centers inside the CsPbClBr₂ SC. We demonstrated multimode lasing in such a CsPbClBr₂/Ag hybrid microcavity at room temperature under the excitation of 325-nm CW laser light. Moreover, we showed ultra-stable emission from the CsPbClBr₂/Ag hybrid microcavity under the continuous excitation of 400-nm femtosecond laser pulses. Our findings indicate the great potential of such hybrid microcavities in making CW-light-pumped perovskite lasers with low threshold and high stability.

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CRediT authorship contribution statement

Shulei Li: Writing – original draft, Validation, Investigation, Funding acquisition, Formal analysis, Data curation. **Zhenxu Lin:** Writing – original draft, Validation, Methodology, Formal analysis, Data curation. **Quantong Deng:** Investigation. **Fu Deng:** Formal analysis. **Mingcheng Panmai:** Formal analysis. **Junyong Chen:** Formal analysis. **Yuheng Mao:** Formal analysis. **Shimei Liu:** Data curation. **Jun Dai:** Data curation. **Yunbao Zheng:** Investigation. **Rui Huang:** Supervision, Investigation, Funding acquisition. **Sheng Lan:** Writing – review & editing, Resources, Project administration, Methodology, Funding acquisition,

Formal analysis.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

No data was used for the research described in the article.

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